

AN EFFICIENT SYNTHESIS OF SUGAR PYRUVIC ACID ACETALS

Peter M. Collins*, Andrew C. McKinnon, and Ajay Manro

Department of Chemistry, Birkbeck College, London WC1E 7HX, UK.

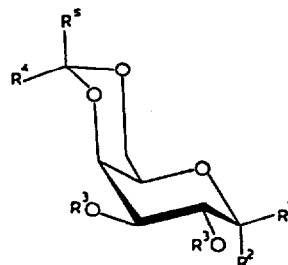
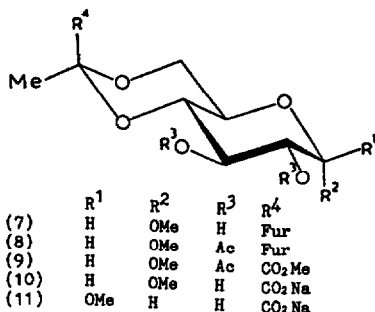
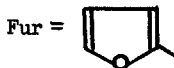
ABSTRACT - 4,6-O-(Carboxyethylidene)-glycopyranosides (e.g. 10) may be prepared by oxidation of the corresponding furan-2-ylethylidene derivatives (e.g. 7) which in turn are readily produced by acetal exchange between the 4,6-hydroxy groups of glycopyranosides.

Sugar pyruvate acetals are important¹ since they appear in many biologically significant polysaccharides and may constitute a vital part of the immunodominant region. Their preparation by the usual methods in which ketone carbonyls are condensed directly with the diols of sugars do not work, probably because the electron withdrawing carboxy group of the pyruvic acid derivative discourages ring closure onto the first formed hemiacetal. The modified reaction in which diols are exchanged with dimethyl acetals also terminates at the noncyclic mixed acetal stage. In Gorin's² pioneering work acetoxymethyl was condensed, albeit in very low yield, with the 4,6-diol of a sugar, with subsequent conversion of the acetoxymethyl group into a carboxylate. One new approach³ to this problem has been to use silylated hydroxy compounds.

We on the other hand have investigated the use of methyl aryl and furan-2-yl ketone dimethyl acetals such as (1)-(6), which all exchange with pairs of sugar hydroxyls to give cyclic acetals that in principle may be oxidized with ruthenium tetroxide⁴ or ozone⁵ to give pyruvic acid acetals. However, in model experiments with these acetals RuO₄ oxidation using the conditions described below only occurred at a useful rate with those derived from the first two, with the former appearing to be the most satisfactory. Oxidation with O₃ was less satisfactory, acetals derived from (1), for example, gave very complex mixtures of products. The general procedure is illustrated by the preparation of methyl 4,6-O-S-(1-carboxyethylidene)- α -D-glucopyranoside (10). Thus treatment⁶ of methyl α -D-glucopyranoside with (1) gave the 2,3-dihydroxy 4,6-O-(furan-2-ylethylidene) β -isomer (7)⁷ which, after chromatography, was acetylated at the 2,3-positions to give pure (8) in 70% overall yield. Oxidation of (8) (10 g), by a method similar to that of Sharpless,⁴ followed by esterification gave, in 74% yield, the acetylated acetal ester (9) which was readily saponified to the sodium salt (10)⁸ in 95% yield. The β -isomer (11)⁹ was identically prepared in similar yield.



- (1) R = Fur
 (2) R = 3,4-(MeO)₂C₆H₃
 (3) R = C₆H₅
 (4) R = 4-BrC₆H₄
 (5) R = 3-(CN)C₆H₄
 (6) R = 4-(CN)C₆H₄



	R ¹	R ²	R ³	R ⁴	R ⁵
(12)	H	OMe	Ac	Me	Fur
(13)	H	OMe	Ac	Fur	Me
(14)	OMe	H	Ac	Fur	Me
(15)	OMe	H	Ac	Me	Fur
(16)	OBn	H	Ac	Me	Fur
(17)	OBn	H	Ac	Fur	Me
(18)	H	OMe	H	Me	CO ₂ Na
(19)	H	OMe	H	CO ₂ Na	Me
(20)	OMe	H	H	CO ₂ Na	Me
(21)	OMe	H	H	Me	CO ₂ Na
(22)	OBn	H	H	Me	CO ₂ Na
(23)	OBn	H	H	CO ₂ Na	Me

To effect exclusive 4,6-acetalation with galactopyranosides their 2,3-diacetates were used in exchange reactions with (1). The 4,6-O-(furan-2-ylethylidene) 2,3-diacetates (12)-(17) were formed in good yield (75-90%) but as R-(equatorial Me) and S-stereoisomers in approximately a 2:1 ratio. Oxidation of the chromatographically pure R- and S- isomers of the methyl and benzyl 4,6-O-(furan-2-ylethylidene) galactosides (12-17) and subsequent methylation and saponification gave the sodium pyruvate acetals (18-20)⁹ (21)¹⁰ and (22 and 23)¹⁰ in high yields.

Acknowledgement We thank the Wellcome Trust for generous support.

References and Notes

- L. Kenne and B. Lindberg in *Polysaccharides*, A.P. London, 1983, pp. 287-363.
- P.A.J. Gorin and T. Ishikawa, *Canad. J. Chem.*, 1967, **45**, 521.
- T. Tsunoda, M. Suzuki, and R. Nogoya, *Tetrahedron Lett.*, 1980, **21**, 1357; H. Hashimoto, K. Hiruma, J. Tamura, *Carbohydr. Res.*, 1988, **177**, C9.
- J.A. Caputo and R. Fuchs, *Tetrahedron Lett.*, 1967, 4729; P.H.J. Carlsen, T. Katsuki, V.S. Martin and K.B. Sharpless, *J. Org. Chem.*, 1981, **46**, 3963.
- A.H. Haines, *Methods for the oxidation of organic compounds*, A.P., 1985, p. 186.
- M.E. Evans, F.W. Parrish, and L. Long Jr., *Carbohydr. Res.*, 1967, **3**, 453.
- All new compounds gave satisfactory ¹H- and ¹³C-nmr spectra and ms or chemically determined elemental compositions.
- Known compounds had physical constants and nmr spectra identical to those reported.
- P.J. Garegg, P. Jansson, B. Lindberg, F. Lindh, J. Lonngren, I. Kvarnstrom, and W. Nimnich, *Carbohydr. Res.*, 1980, **78**, 127.
- Values for δ_C (50 MHz) and where relevant δ_H (200 MHz) for the anomeric position, Me, C=O, and O₂CR₂ in compounds: (21) 106.2, proton signal obscured; 18.8, 1.60; 179.3; 100.8; (22) 104.0, 4.46; 18.0, 1.60; 179.2; 100.8; (23) 104.0, 4.47; 28.0, 1.48; 178.9; 103.6.

(Received in UK 30 December 1988)